Student Name	Date

Oklahoma Physical Science Framework

Activity (Part 1: absorbed - magnetic)

For each word listed in the left column, find a sentence in the text containing that word. Copy the complete sentence in the space provided. <u>Underline</u> the vocabulary word in your sentence.

Example

accelerations	An applied force will cause a change in the types of energy in the system, therefore the energy found in waves and the interaction of waves and their
	environment will cause <u>accelerations</u> .

Word List

absorbed	
Acceleration	
accelerations	
accomplish	
according	
accounted	
acting	

addition		
advantages		
amount		
Analyze		
apart		
applied		
Apply		
arranged		
associated		
atom		
Atomic		
atoms	 	

attractive		
available		
balance		
based		
behaviors		
Bioshpere		
bond		
building		
calculate		
capable		
capture		
Cause		

Student Name	Date
cells	
changes	
charge	
chemical	
claim	
cloud	
collision	
combined	
component	
compounds	
computational	
concentration	

conduct		
conduction		
Conservation		
conserved		
constraints		
Construct		
continue		
convert		
converted		
Create		
current		
damage		

data		
describes		
designed		
determine		
Develop		
device		
different		
digital		
direction		
disadvantages		
disordered		
distribution		

Student Name	 Date	
effects		
electric		
electrical		
electromagnetic		
electron		
element		
empty		
energy		
engineering		
environment		
Equilibrium		
Evaluate		

evidence		
explanation		
exposure		
field		
fields		
flows		
force		
form		
formation		
frequencies		
given		
goals		

group		
happen		
ideas		
identify		
illustrate		
impact		
important		
increase		
increases		
indicates		
influenced		
information		

integral		
interaction		
investigation		
kinetic		
knowledge		
law		
level		
living		
long		
machines		
macroscopic		
magnetic		

Activity (Part 2:	mass – waves)	
For each word listed in the left column, find a sentence in the text containing that word. Copy the		
complete sentence in	the space provided. <u>Underline</u> the vocabulary word in your sentence.	
Example		
accelerations	An applied force will cause a change in the types of energy in the system, therefore the energy found in waves and the interaction of waves and their environment will cause <u>accelerations</u> .	
mass		
materials		
mathematical		
matter		
minimize		
models		
molecules		
Momentum		
motion		

Student Name	 	
Newton		
nuclear		
nucleus		
object		
orbit		
order		
orientation		
original		
outcome		
outermost		
part		
ραιτ		
particles		

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patterns	
period	
Periodic	
Periods	
physical	
placed	
Plan	
position	
possesses	
predict	
present	
principles	

produce		
Properties		
protons		
provide		
pushes		
quantity		
questions		
radiation		
rate		
reached		
reacting		
reaction		

reactions		
reactivity		
rearranged		
Ü		
refine		
na na velica n		
regarding		
relative		
release		
reliability		
remadiney		
remains		
repeating		
representations		
results		
	1	

revise		
scientific		
second		
simple		
size		
society		
space		
speed		
start		
states		
still		
storage		

stored	
substance	
Substance	
support	
system	
3,300	
Table	
temperature	
·	
thermal	
total	
transfer	
transformed	
transmission	
traveling	
trends	
types	

Student Name	Date
uniform	
validity	
vector	
velocity	
wavelength	
waves	

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Text (Passages)

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Atomic Properties and the Periodic Table

The periodic table can be used to identify atomic behaviors/properties and predict the outcome of chemical reactions.

- 1. Elements are placed on the Periodic Table according to repeating patterns of physical and chemical properties, as well as reactivity patterns.
- 2. Atomic size decreases going across a period of the table due to increasing nuclear charge. Atomic size increases down a group of elements due to addition of energy levels.
- 3. All matter is conserved, just broken apart and rearranged to form new molecules/substances.
- 4. A chemical bond is an attractive force not a physical thing at all.
- 5. The electron cloud is a "cloud" because of the motion of the electron in orbit around the nucleus, and mostly made up of empty space.
- 6. Periods on the periodic table are based on the energy levels an atom has.
- 7. The atomic number of an atom indicates the number of protons an atom has which determines what element and therefore the chemical properties it possesses.
- 8. Changing the number of electrons an atom has will change its reactivity with atoms around it.

Properties of Chemical Reactions and Conservation of Energy

Chemical reactions always start and end with the same amount of atoms, though they will be arranged differently. Furthermore; how fast the reaction occurs, and if energy is stored or released is all determined by the collisions of the molecules that make up the chemical reaction. The number of collisions of molecules during a chemical reaction can be influenced by temperature as well as the amount of particles present during a reaction.

- 1. Although the substance changes forms, the atoms are still there. In a closed system you will see no change in mass.
- 2. A chemical reaction will only happen if a collision occurs, however more requirements, such as orientation and available kinetic energy, must also be met.
- 3. All matter is conserved, just broken apart and rearranged to form new molecules/substances.

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Acceleration and Things That Cause Acceleration

Applying a force to an object will cause acceleration. The size of this acceleration is determined by the mass of the object and the size of force applied. An applied force will cause a change in the types of energy in the system, therefore the energy found in waves and the interaction of waves and their environment will cause accelerations.

- 1. If a force acts on an object in the same direction as the direction of its motion, the object's speed will continue to increase while the force is acting.
- 2. Energy can be transferred from one system to another (or from a system to its environment) in different ways: by conduction, mechanically, electrically, or by radiation (electromagnetic waves).
- 3. Energy can be transferred within a system. Regardless of what happens within a system, the total amount of energy in the system remains the same unless energy is added to or released from the system.
- 4. Energy can be transformed (converted) within a system.

Momentum

Momentum is determined by the speed of an object and the direction it is traveling (velocity) of an object and the object's mass. This momentum is conserved as long as there are no new objects added to the system. If a new object is added then the momentum will change in order to maintain a balance in the overall system. Devices can be designed and tested, that will use this balance of forces to minimize the effects of a change in momentum on an object.

- 1. Momentum is a vector quantity.
- 2. Momentum is mass in motion, whereas forces are pushes or pulls applied to an object or mass.
- 3. Momentum is conserved in a collision.

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Defining and Calculating Energy

The energy of a system depends on the motion of the system, as well as the interactions that occur within the system. Energy is always changing from one kind to another, but the total energy of the system is always the same. Energy can take many forms such as motion, sound, light, and heat. The amount of energy available is mathematically calculable, and determines what the system is capable of doing.

- 1. Energy can be transformed (converted) within a system.
- 2. Energy can be transferred from one system to another (or from a system to its environment) in different ways: by conduction, mechanically, electrically, or by radiation (electromagnetic waves).
- 3. Regardless of what happens within a system, the total amount of energy in the system remains the same unless energy is added to or released from the system.
- 4. Regardless of what happens within a system, the total amount of energy in the system remains the same unless energy is added to or released from the system.
- 5. Motion energy (kinetic energy) is associated with the speed and the mass of an object.

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The Use of Energy, Its Conservation, and Equilibrium

Energy can be seen in multiple ways and be used to accomplish goals by building machines that capture and use that energy. These machines will transfer one type of energy to another form until a balance between the amounts of the different forms of energy is reached. Show this by building a machine to accomplish a task.

- 1. Thermal energy of an object is associated with the disordered motions of its atoms or molecules and the number and types of atoms or molecules of which the object is made.
- 2. Energy can be transformed (converted) within a system.
- 3. Energy can be transformed (converted) within a system.
- 4. Energy can be transferred from one system to another (or from a system to its environment) in different ways: by conduction, mechanically, electrically, or by radiation (electromagnetic waves).

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The Use of Electromagnetism and Its Effect on the Bioshpere

Electromagnetic radiation when absorbed can be converted to thermal energy, cause damage to living cells, or even cause materials to release electrons therefore being converted into electrical energy. In addition, the use of electromagnetic waves can be used to send information worldwide and has become an integral part of our society. It is important to determine the full impact of the advantages and disadvantages of our current use of and exposure to electromagnetism.

- 1. Energy can be transferred from one system to another (or from a system to its environment) in different ways: by conduction, mechanically, electrically, or by radiation (electromagnetic waves).
- 2. Only energy is transferred with the wave, the particles always return to their original position.